

RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College affiliated to University of Calcutta)

SECOND YEAR [2017-20]

B.A. /B.Sc. FOURTH SEMESTER (January – June) 2019

Mid-Semester Examination, March 2019

Date : 28/03/2019

PHYSICS (General)

Time : 1pm – 2pm

Paper: IV

Full Marks: 25

Group - A

(2×5)

Answer **any two** of the following questions:

1. a) Explain the formation of the "depletion region" in an open circuited PN junction. [2]
b) What causes majority carriers to flow at the moment when a P region and an N region are brought together? How long this flow will continue? Explain. [3]
2. a) Draw the circuit diagram of a half-wave rectifier. [1]
b) Prove that the ripple factor of a half wave rectifier is 1.21. [4]
3. Draw the circuit diagram of a full wave rectifier using
 - a) Centre tap
 - b) Bridge connectionExplain the working of each. Why bridge connection is preferred than centre tap? [2+2+1]
4. Draw the circuit diagram of a voltage regulator using a zener diode. Explain its working. [2+3]

Group - B

(3×5)

Answer **any three** of the following questions:

5. a) What is packing fraction of atom. [1]
b) Draw a curve between packing fraction and mass number and discuss the nature of the curve. [2+1]
c) What is Bragg's law of diffraction? [1]
6. a) Find the Q-value in MeV for the following nuclear reaction: [2]
$${}_5\text{B}^{10} + {}_1\text{H}^2 \rightarrow {}_5\text{B}^{11} + {}_1\text{H}^1$$

mass of $\text{B}^{10} = 10.0165 \text{ a.m.u}$
 $\text{B}^{11} = 11.01286 \text{ a.m.u}$
 $\text{H}^2 = 2.01472 \text{ a.m.u}$
 $\text{H}^1 = 1.0082 \text{ a.m.u}$

b) 1gm of radium ejects 3.5×10^{10} \propto particles per second. Calculate the half-life of radium, given atomic mass of radium = 226. [3]
7. How energy released in nuclear fusion process. [5]

8. What is Compton's scattering?

Derive the relation for Compton shift in wave length of the incident photon.

[1+4]

9. a) Write down the postulates of Bohr's atomic model. What are the limitations of this model.

b) In a hydrogen like atom, electrons make transitions from energy level with quantum number 'n' to another quantum number 'n-1'. If $n \gg 1$, show that frequency of radiation will be proportional to $\frac{1}{n^3}$.

[3+2]

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